

Received: November 16, 1989; accepted February 1, 1990

PRELIMINARY NOTE

The Synthesis from 2,5-Dichloro-1,1,1,6,6,6-hexafluoro-3,4-diazahexa-2,4-diene of 4H-3,5-Bis(trifluoromethyl)-1,2,4-triazole and some 1- and 4-Substituted Derivatives

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SUMMARY

Treatment of the title azine (**3**) with an excess of ammonia or ethyl glycinate affords the corresponding diaminoazines (**4a**) and (**4b**), which undergo thermal cyclisation to give ammonium 3,5-bis(trifluoromethyl)-1,2,4-triazolate (**6a**) [from which the parent 4H-1,2,4-triazole (**1a**) can be liberated by aqueous acid treatment] and 4-carboethoxymethyl-3,5-bis(trifluoromethyl)-1,2,4-triazole (**1e**), respectively in high yield. Reaction of the 4H-1,2,4-triazole (**1a**) with ethyl propynoate and of its sodium salt (**6b**) with ethyl bromoethanoate and pentafluoropyridine yield the acrylate (**7a**)-, carboethoxymethyl (**7b**)- and fluoropyridyl (**9-11**)- derivatives of 1H-3,5-bis(trifluoromethyl)-1,2,4-triazole with competing attack on the pentafluoropyridine ring taking place at the 2- and 4-positions.

4H-3,5-Bis(trifluoromethyl)-1,2,4-triazole (**1a**) and its 4-methyl derivative (**1b**) were first synthesised from the oxadiazole (**2**) via reaction with ammonia and methylamine, respectively [1], and this route has been extended recently to the preparation of a wide range of 4-alkyl-, 4-alkenyl- and 4-aryl-1,2,4-triazoles [2].

In a previous communication we reported the reactions of 2,5-dichloro-1,1,1,6,6,6-hexafluoro-3,4-diazahexa-2,4-diene (**3**) with a variety of nucleophiles including primary amines and described the preparation of 4-phenyl (**1c**)- and 4-(2,6-dichloropyridyl)methyl (**1d**)- triazoles [3].

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We now report the synthesis of the parent 4H-1,2,4-triazole (**1a**) and its 4-carboethoxymethyl derivative (**1e**) by the same route from azine (**3**) and the conversion of triazole (**1a**) into certain derivatives of 1H-3,5-bis(trifluoromethyl)-1,2,4-triazole.

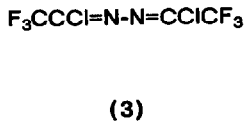
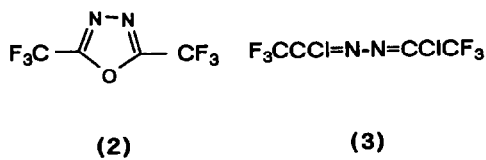
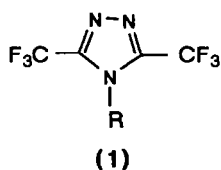
Reaction of dichloroazine (**3**) with an excess of aqueous ammonia or ethyl glycinate in diethyl ether at room temperature afforded the diaminoazine (**4a**) (95%) and a mixture of the monoglycinatoazine (**5**) (16%) and diglycinatoazine (**4b**) (75%) respectively. Thermolysis of the diaminoazines (**4a**) and (**4b**) in a sealed tube *in vacuo* at ca. 150 °C gave ammonium 3,5-bis(trifluoromethyl)-1,2,4-triazolate (**6a**) (98%) and 4-carboethoxymethyl-3,5-bis(trifluoromethyl)-1,2,4-triazole (**1e**) (82%) respectively. Treatment of the salt (**6a**) with 4M hydrochloric acid yielded the 4H-triazole (**1a**) (75%).

Thus, the reaction of the dichloroazine (**3**) with primary amines followed by thermal cyclisation of the resulting diaminoazines provides a satisfactory alternative general route to that from the oxadiazole (**2**) [2] for the preparation of 4-substituted-3,5-bis(trifluoromethyl)-1,2,4-triazoles.

The reaction of triazole (**1a**) with ethyl propynoate was investigated to determine whether the triazole was an effective nucleophile for Michael addition and whether attack on the substrate would involve N-1 or N-4. A mixture of (**1a**) and ethyl propynoate heated *in vacuo* in a sealed tube at 120 °C (14 days) gave the Michael adduct, ethyl 3-[1H-3,5-bis(trifluoromethyl)-1,2,4-triazolino]acrylate (**7a**) (95%), as a mixture of the (E)- and (Z)-isomers (ratio 46:49) which were separated by dry column flash chromatography (DCFC) (eluant:n-C₆H₁₄/CHCl₃ 1:1 v/v). This reaction occurred exclusively at N-1 of the triazole ring possibly due to N-4 being β to two strongly electron-withdrawing CF₃ groups which reduce the availability of the lone pair on N-4 and sterically hinder reaction at this position; N-1 is β to only one CF₃ group.. It has been reported that reaction of the 4H-1,2,4-triazole (**8**) with N-chloromethylacetanilides afforded 4-substituted triazoles [4].

The resonance-stabilised salts of 3,5-disubstituted-1,2,4-triazoles undergo reaction at N-1 because of the greater nucleophilicity of the N-N linkage relative to N-4 [5]. Therefore, the reactions of the sodium salt (**6b**) [prepared by treatment of triazole (**1a**) with sodium hydride in diethyl ether or THF] with ethyl bromoethanoate [a potential route to the 1-substituted isomer (**7b**) of triazole (**1e**)] and with pentafluoropyridine (to determine if nucleophilic attack at the 2-position of the pyridine ring could compete with 'normal' attack at the 4-position) were investigated.

Treatment of salt (**6b**) with ethyl bromoethanoate at room temperature in diethyl ether gave 1-carboethoxymethyl-3,5-bis(trifluoromethyl)-1,2,4-triazole (**7b**) (87%), while with pentafluoropyridine in THF heated under reflux a mixture was



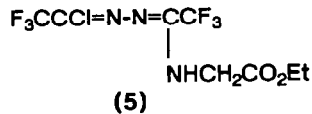
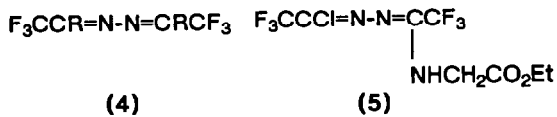
a, R = H

b, R = Me

c, R = Ph

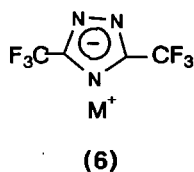
d, R = CH₂-

e, R = CH₂CO₂Et



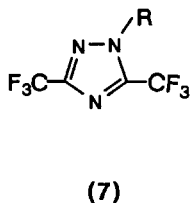
a, R = NH₂

b, R = NHCH₂CO₂Et



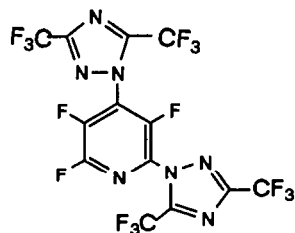
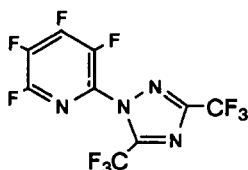
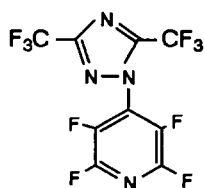
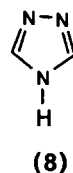
a, M = NH₄

b, M = Na



a, R = CH=CHCO₂Et

b, R = CH₂CO₂Et



formed from which the major products were separated by DCFC (eluant $n\text{-C}_6\text{H}_{14}$) and identified as the 4-triazolinotetrafluoropyridine (9) (19%), the 2-triazolinotetrafluoropyridine (10) (5%) and the 2,4-bis(triazolino)trifluoropyridine (11) (6%) all formed via attack by N-1 of the triazolate on the pyridine ring.

It has been reported recently [6] that treatment of the sodium or potassium salts of the oximates $\text{RR}'\text{C}=\text{NOH}$ ($\text{R}=\text{R}'=\text{Me}$; $\text{R}=\text{R}'=\text{Ph}$; $\text{R}=\text{Me}$, $\text{R}'=\text{Ph}$) and the lithium salts of the hydrazones $\text{Ph}_2\text{C}=\text{NNHR}$ ($\text{R}=\text{H}$ or Ph) with pentafluoropyridine afforded mixtures of the corresponding 2- and 4-substituted tetrafluoropyridines in contrast to reaction with other nucleophiles which involved exclusive initial attack in the 4-position [7]. The present reaction involving the salt (6b) thus extends the examples of nucleophiles which attack the 2-position of pentafluoropyridine in competition with the 4-position.

Further reactions of triazole (1a) and its sodium salt (6b) are under investigation and will be reported in due course in a full paper.

All products possessed satisfactory elemental compositions and their structures were determined spectroscopically [i.r., n.m.r. (^1H , ^{13}C , ^{19}F) and mass].

We thank the committee of Vice Chancellors and Principals for an O.R.S. award (to M. A-G.).

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